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Highly efficient amphiphilic cleavage of γ -iodo carbonyl substrates with aluminum tris(2,6-diphenylphenoxide)/*t*-BuLi system

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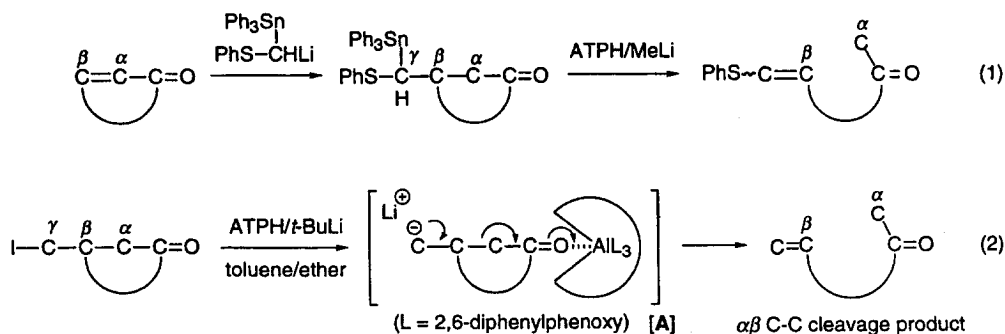
Abstract

A conceptually new amphiphilic cleavage of the $\alpha\beta$ C–C bonds of γ -iodo carbonyl substrates has been realized by the effective use of a combined Lewis acid/base system consisting of aluminum tris(2,6-diphenylphenoxide) (ATPH)/*t*-BuLi. This new amphiphilic bond cleavage reaction can be applied to a wide variety of γ -iodo carbonyl substrates and therefore serves as a highly efficient and general route to both cyclic and acyclic unsaturated carbonyl compounds. © 1999 Elsevier Science Ltd. All rights reserved.

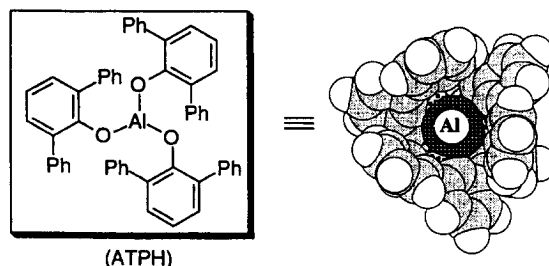
Keywords: aluminum; aluminum compounds; cleavage reaction; lithium; lithium compounds; carbonyl compounds.

We recently disclosed a novel fragmentation reaction of enones via the successive conjugate addition–cleavage sequence by the use of $\text{Ph}_3\text{Sn}(\text{PhS})\text{CHLi}$ followed by the aluminum tris(2,6-diphenylphenoxide) (ATPH)/MeLi system (Eq. 1).^{1,2} This new synthetic transformation, however, has one drawback on the limited use of starting α,β -unsaturated carbonyl substrates, and indeed α,β -unsaturated aldehydes and esters are not employable for the present amphiphilic cleavage reaction.¹ This is mainly due to the use of sulfur- or selenium-stabilized carbanions. Accordingly, we were interested in the possibility of utilizing non-stabilized pure carbanions in order to effect the facile $\alpha\beta$ C–C bond cleavage by enhancing the push effect of the carbanion parts. Here we wish to report our successful results on highly efficient, amphiphilic cleavage of the $\alpha\beta$ C–C bonds of γ -iodo carbonyl substrates to generate pure carbanions by treatment with *t*-BuLi via the facile metal–halogen exchange (Eq. 2), thereby allowing hitherto difficult cleavage of $\alpha\beta$ C–C bonds of aldehydes and esters in the presence of ATPH.³

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Scheme 1.

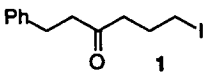
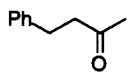
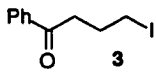

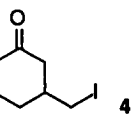
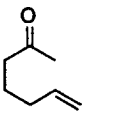
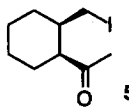
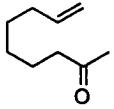
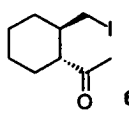
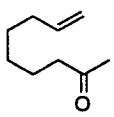
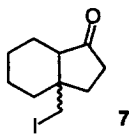
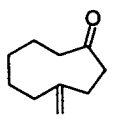
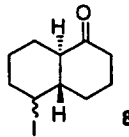
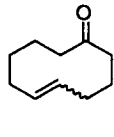
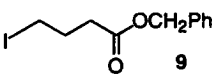
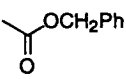
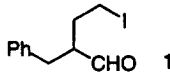
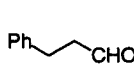


The starting γ -iodo carbonyl substrates can be readily prepared by the following synthetic methods: (A) γ -iodo ketones: iodination of γ -hydroxyketones with $\text{Ph}_3\text{P}/\text{Py}/\text{I}_2$; (B) γ -iodo ketones: cyclopropanation of enones with dimethyloxosulfonium methylide,⁴ and subsequent ring opening of the resulting cyclopropyl ketones with Me_3SiI ; (C) γ -iodo esters: cleavage of γ -butyrolactone derivatives with MeSiI in the presence of alcohol; and (D) γ -iodo aldehydes: reduction of γ -iodo esters with DIBAH.

Initial complexation of 6-iodo-1-phenyl-3-hexanone (**1**) with ATPH (1.1 equiv.) in toluene/ether (volume ratio=1:1) at -78°C and subsequent addition of *t*-BuLi (2 equiv.) in pentane cleanly afforded 4-phenyl-2-butanone (**2**) as an $\alpha\beta$ C-C bond cleavage product in almost quantitative yield (entry 1 in Table 1). The clean and rapid halogen-lithium exchange takes place in a toluene-ether solvent at -78°C , while in only toluene solvent the lithiation proceeds reluctantly. The simultaneous push-pull effect of the intermediary γ -lithio carbonyl substrates by complexation with Lewis acidic ATPH as a carbonyl protector seems to be crucial for effecting the smooth $\alpha\beta$ C-C bond cleavage reaction as shown in [A] of Scheme 1. Indeed, in the absence of ATPH, treatment of γ -iodo ketone **1** with *t*-BuLi gave deteriorated reaction mixtures without formation of the desired **2**. Attempted use of ordinary Lewis acids such as $\text{BF}_3 \cdot \text{OEt}_2$ and TiCl_4 in place of ATPH gave unsatisfactory results mainly due to the facile attack of *t*-BuLi to the Lewis acids.

Other selected examples, which are shown in Table 1, concerning the novel fragmentation reaction of γ -iodo carbonyl substrates with the ATPH/*t*-BuLi system, clearly indicate the effectiveness of our approach.⁵ Although aromatic ketones on treatment with ATPH are reported to give the corresponding 1,6-adduct as a nucleophilic aromatic substitution product,^{3c} acetophenone was obtainable exclusively from γ -iodobutyrophenone (**3**) without concomitant formation of Michael addition products on the benzene ring under these reaction conditions (entry 2). β -Iodomethylcyclohexanone (**4**), *cis*-(2-iodomethyl)cyclohexyl methyl ketone (**5**), and its *trans*-isomer **6** experienced clean ring-cleavage by treatment with the ATPH/*t*-BuLi system to furnish the corresponding alkenyl methyl ketones in good yields (entries 3-5). Bicyclic ketone **7** possessing an iodomethyl moiety underwent smooth ring expansion to furnish a large-membered cyclic ketone with *exo* double bond (entry 6), which represents

Table 1
 Selective cleavage of γ -iodo carbonyl substrates with ATPH/*t*-BuLi System^a

entry	substrate ^b	cleavage product	% yield ^c
1			>99% ^d
2			>99%
3			80% ^d
4			93%
5			70% ^e
6			85%
7			50% ^f
8			84%
9			58% ^e

^a The reaction was carried out with ATPH (1.1 equiv) and *t*-BuLi (2 equiv) in toluene/ether (volume ratio = 1:1) at -78 °C for 20 min. ^b Prepared according to the method A (entries 1, 2, and 7), B (entries 3-6), C (entry 8), and D (entry 9), respectively. ^c Isolated yield. ^d Isolated yield after conversion of fragmentation product to the corresponding alcohol by DIBAH reduction in toluene at -78 °C. ^e Cyclobutanol derivatives, which arised from intramolecular nucleophilic addition of the initially generated carbanion to carbonyl moiety, were also obtained as side products (10–20% yields). ^f *E/Z* ratio of the product was determined to be 1:1.6 by ¹H NMR analysis.

an attractive method for the construction of otherwise difficult nine-membered carbocycles. Formation of ten-membered carbocycles also appears feasible starting from a *secondary* iodo decalone **8** (entry 7). γ -Iodo ester **9** and aldehyde **10** were also susceptible to the new amphiphilic bond cleavage reaction, producing the corresponding fragmentation products (entries 8 and 9).

The new amphiphilic cleavage of γ -iodo carbonyl substrates described herein serves as a highly efficient and general route to a variety of both cyclic and acyclic unsaturated carbonyl compounds, and, therefore, further expands utility in selective organic synthesis.

Acknowledgements

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5. A typical experimental procedure for cleavage of γ -iodo ketone **1** with ATPH/*t*-BuLi system follows: A solution of 2,6-diphenylphenol (406 mg, 1.65 mmol) in toluene (4.5 mL) was carefully degassed and a 1 M hexane solution of Me₃Al (0.55 mL, 0.55 mmol) was added at room temperature under argon. Methane gas evolved immediately. The resulting yellow solution was stirred for 30 min and used without purification. After the addition of 6-iodo-1-phenyl-3-hexanone (**1**) (151 mg, 0.5 mmol) in toluene (0.5 mL) and diethyl ether (5 mL) at –78°C, a 1.6 M pentane solution of *t*-BuLi (625 μ L, 1 mmol) was added dropwise. The solution was stirred at –78°C for 20 min and then poured into 1 N HCl solution. Extractive workup was performed with ether and the organic extracts were dried over Na₂SO₄. Evaporation of solvents and purification of the residue by column chromatography on silica gel (dichloromethane/hexane=1:3 to ethyl acetate/hexane=1:9 as eluant) gave 4-phenyl-2-butanone (**2**) as a colorless oil (73.7 mg, 0.497 mmol; 99% yield).